



Code No. : **9273/AB**

FACULTY OF SCIENCE
M.Sc. II Semester Examination, May/June 2012
CHEMISTRY

Paper – IV : Computers in Chemistry and Spectroscopy

Time : 3 Hours]

[Max. Marks : 80

Note : *This question paper consists of two parts. Part A is Computers in Chemistry. Part B is Spectroscopy. Write Part A and Part B in **separate** answer books. Answer **all** questions.*

PART – A
(Computers in Chemistry)

SECTION – A

(1×8=8 Marks)

1. a) Write a brief note on input and output devices.
- b) What are the steps involved in testing a source program in C ?

SECTION – B

(1×12=12 Marks)

2. a) Give the syntax for printf(), scanf () and puts with examples.
- b) Write the listings of a C program for roots of a quadratic equation.

OR

- c) Define an array. Write the syntax for a two dimensional array.
- d) Write the listings of a C program for first order rate constant from kinetic data.

PART – B
(Spectroscopy)

SECTION – A

(3×8=24 Marks)

1. a) Write a note on lanthanide shift reagents.
- b) How nmr spectrum can be simplified by deuterium exchange ?



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2. a) Explain nitrogen rule with examples.
b) What are meta stable peaks ? How can they be identified ?
3. a) What are the characteristic fragmentation pattern of cyclo alkanols ?
b) What is ortho effect ?

SECTION – B

(3×12=36 Marks)

4. a) Explain how nmr spectra can be simplified by
 - i) increasing the field strength and
 - ii) double resonance.
b) What is NOE ? Give examples.

OR

- c) Sketch the ^1H and ^{13}C nmr of $^{13}\text{CH}_4$. Explain its splitting pattern.
 - d) What are the applications of ^{19}F nmr spectra ?
5. a) Describe the working of electron spray ionisation mass spectrometer.
b) What is the principle involved in LC - MS interface ? What are its advantages and disadvantages ?

OR

- c) Explain the principle involved in quadrupole mass analyzer.
 - d) What is FAB technique ? What type of molecules can be analyzed by this technique ?
6. a) An organic compound ($\text{C}_8\text{H}_8\text{O}_2$) gave peaks at m/z 136, 105, 77 and 51. Identify the compound and indicate the fragmentation pattern.
b) Explain the fragmentation pattern of aliphatic amines with suitable example.

OR

- c) Explain McLafferty and double McLafferty rearrangements with suitable examples.
- d) Explain the mass spectral fragmentation of $\text{ReBr}(\text{CO})_5$.